

Goyal Brothers Prakashan Chemistry Lab Manual Class

Anil Sharma, Prashant Sharma, Goyal Brothers Prakashan

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Solutions to Learning Elementary Chemistry for Class 7 Dr. R. Goel, 2021-01-01

Practical/Laboratory Manual Chemistry Class - XI Er. Meera Goyal, 2021-05-29

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 2. To bend the glass rod at an angle,
 3. To draw a glass jet from a glass tube
 4. To bore a cork and fit a glass tube into itViva-Voce
2. Characterisation and Purification of Chemical Substances
 1. To determine the melting point of the given unknown organic compound and its identification (simple laboratory technique) Viva-Voce
 2. To determine the boiling point of a given liquid when available in small quantity (simple laboratory method) Viva-Voce
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 4. To prepare the pure crystals of copper sulphate from the given crude sample
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Core Laboratory Manual of Physics for Class XI Anil Sharma, Prashant Sharma, Goyal Brothers Prakashan, 2020-04-16
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given liquid when available in small quantity (simple laboratory method), 3. To prepare crystals of pure potash alum $[K_2SO_4 \cdot Al_2(SO_4)_3 \cdot 24H_2O]$ from the given impure sample, 4. To prepare the pure crystals of copper sulphate from the given crude sample, 5. To prepare pure crystals of benzoic acid from a given impure sample. (C) Measurement of pH Values 1. To determine the pH value of vegetable juices, fruit juices, tap water and washing soda by using universal pH paper, 2. To determine and compare the pH values of solutions of strong acid (HCl) and weak acid (CH₃COOH) of same concentration, 3. To study the pH change in the titration of strong base Vs. strong acid by using universal indicator paper, 4. To study the pH change by common ion (CH₃COO⁻ ion) in case of weak acid (CH₃COOH), 5. To determine the change in pH value of weak base (NH₄OH) in presence of a common ion (NH₄⁺), (D) Chemical Equilibrium 1. To study the shift in equilibrium between ferric ions and thiocyanate ions by changing the concentrations of either of the ions, 2. To study the shift in equilibrium between $[Co(H_2O)_6]^{2+}$ and Cl⁻ ions by changing the concentrations of either of the ions, (E) Quantitative Analysis 1. To prepare M/10 oxalic acid solution by direct weighing method, 2. To prepare M/10 solution of sodium carbonate by direct weighing method, 3. To determine the strength of given solution of sodium hydroxide by titrating it against N/10 or M/20 solution of oxalic acid, 4. To determine the strength of a given solution of hydrochloric acid by titrating it against a standard N/10 or M/20 sodium carbonate solution, (F) Qualitative Analysis 1. Analysis of Anions, 2. Analysis of Cations (G) Detection of Elements in Organic Compounds 1. To detect the presence of nitrogen, sulphur and halogens in a given organic compound by Lassaigne's test, 2. To detect the presence of nitrogen, sulphur and halogens in the given organic compound sample number by Lassaigne's test INVESTIGATORY PROJECTS (A) Checking of Bacterial Contamination in Water 1. To check the bacterial contamination in drinking water by testing sulphide ions (B) Methods of Water Purification 1. To purify water from suspended impurities by using sedimentation, 2. To purify water by boiling, 3. To purify water by distillation method, 4. To purify water by reverse osmosis technique. 5. To purify water by GAC method, 6. To purify water by bleach treatment, 7. To purify water by oxidising agent, 8. To purify water by ozone treatment method. (C) Water Analysis 1. To test the hardness of different water samples. (D) Foaming Capacity of Various Soaps 1. To compare the foaming capacity of different washing soaps, 2. To study the effect of addition of sodium carbonate on foaming capacity of washing soap (E) Tea Analysis 1. To study the acidity of different samples of tea leaves (tea) by using pH paper (F) Analysis of Fruits and Vegetable Juices 1. To analyse the fruit and vegetable juices for the constituent present in them (G) Rate of Evaporation 1. To study the rate of evaporation of different liquids (H) Effect of Acids and Bases on Tensile Strength of Fibres 1. To compare the tensile strength of natural fibres and synthetic fibres, 2. To study the effect of acids and bases on tensile strength of different fibres. Log & Antilog Table

Practical/Laboratory Manual Chemistry Class XII based on NCERT guidelines by Dr. S. C. Rastogi, Er. Meera Goyal Dr. S. C. Rastogi, Er. Meera Goyal, 2020-06-22 A. Surface Chemistry 1. To prepare colloidal solution (sol) of starch, 2. To prepare a colloidal solution of egg albumin 3. To prepare colloidal solution of gum, 4. To prepare colloidal solution of

aluminium hydroxide $[\text{Al}(\text{OH})_3]$, 5. To prepare colloidal solution of ferric hydroxide $[\text{Fe}(\text{OH})_3]$, 6. To prepare colloidal solution of arsenious sulphide $[\text{As}_2\text{S}_3]$, 7. To purify a freshly prepared sol by dialysis, 8. To compare the effectiveness of different common oils (Castor oil, cotton seed oil, coconut oil, kerosene oil, mustard oil) in forming emulsions.

Viva-Voce B. Chemical Kinetics

- To study the effect of concentration on the rate of reaction between sodium thiosulphate and hydrochloric acid,
- To study the effect of temperature on the rate of reaction between sodium thiosulphate and hydrochloric acid,
- To study the rate of reaction of iodide ions with hydrogen peroxide at different concentrations of iodide ions,
- To study the rate of reaction between potassium iodate (KIO_3) and sodium sulphite (Na_2SO_3) using starch solution as indicator

Viva-Voce C. Thermochemistry

- Determine the enthalpy of dissolution of copper sulphate ($\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$) in water at Room temperature,
- To determine the enthalpy of neutralization of the reaction between HCl and NaOH ,
- To determine enthalpy change during the interaction between acetone and chloroform

Viva-Voce D. Electrochemistry

- To study the variation of cell potential in $\text{Zn}|\text{Zn}^{2+}||\text{Cu}^{2+}|\text{Cu}$, with change in concentration of electrolytes (CuSO_4 or ZnSO_4) at room temperature

Viva-Voce E. Chromatography

- To separate the coloured components (pigment) present in the given extract of leaves and flowers by ascending paper chromatography and find their R_f values,
- To separate the coloured components present in the mixture of red and blue inks by ascending paper chromatography and find their R_f values,
- To separate Co^{2+} and Ni^{2+} ions present in the given mixture by using ascending paper chromatography and determine their R_f values

Viva-Voce F. Preparation of Inorganic Compounds

- Preparation of double salt of ferrous ammonium sulphate (Mohr's salt) from ferrous sulphate and ammonium sulphate,
- To prepare a pure sample of potash alum (fitkari),
- Preparation of crystals of potassium ferric oxalate or potassium trioxalato ferrate (III)

Viva-Voce G. Preparation of Organic Compounds

- Preparation of iodoform from ethyl alcohol or acetone,
- Preparation of acetanilide in laboratory,
- Preparation of *b*-Naphthol aniline dye,
- To prepare a pure sample of dibenzalacetone,
- To prepare a pure sample of *p*-nitro acetanilide

Viva-Voce H. Tests for the Functional Groups Present in Organic Compounds

Viva-Voce I. Study of Carbohydrates, Fats and Proteins

- To study simple reactions of carbohydrate,
- To study simple reactions of fats,
- To study simple reactions of proteins,
- To investigate presence of carbohydrates, fats and proteins in food stuffs

Viva-Voce J. Volumetric Analysis

- To prepare 250 ml of $M/10$ solution of oxalic acid,
- To prepare 250 ml of $M/10$ solution of ferrous ammonium sulphate,
- Prepare $M/20$ solution of oxalic acid, with its help find out the molarity and strength of the given solution of potassium permanganate,
- Prepare $M/20$ solution of Mohr's salt, using this solution determine the molarity and strength of potassium permanganate solution

Viva-Voce K. Qualitative Analysis

Viva-Voce INVESTIGATORY PROJECTS

- To study the presence of oxalate ions in guava fruit at different stages of ripening.
- To study the quantity of casein present in different samples of milk.
- Preparation of soyabean milk and its comparison with natural milk with respect to curd formation, effect of temperature etc.
- To study the effect of potassium bisulphite as food preservative at various concentrations.
- To study the digestion of starch by salivary amylase

and the effect of pH and temperature on it. 6. To study and compare the rate of fermentation of the following materials—wheat flour, gram flour, potato juice and carrot juice. 7. To extract essential oils present in saunf (aniseed), ajwain (corum), illaichi (cardomom). 8. To detect the presence of adulteration in fat, oil and butter, 9. To investigate the presence of NO₂- in brinjal.

Chemistry Lab Manual Neena Sinha, R Rangarajan, R P Manchanda, R K Gupta, Rajesh Kumar, Lab Manual

Physical Chemistry Laboratory Manual Ramesh Kumari, Amirtha Anand, 2018-10-05 This book covers the latest syllabus of CBCS pattern of Delhi and other universities for both B.Sc. Programme and Honours courses. A large number of Physical Chemistry, Environmental Chemistry, Nanoscience, Polymer Chemistry and Analytical Chemistry experiments have been covered using interdisciplinary and innovative methods. The contents include some fundamental chemical concepts, measurement of surface tension and viscosity, colorimetry, determination of order of a reaction, heterogeneous equilibria, adsorption on solid surfaces, thermochemical measurements, conductometric and potentiometric measurements, pH metry, environmental parameter analysis, etc. Wherever possible, two or more methods are given. So the teachers and students will have a choice to make depending on the availability of chemicals, apparatus, instruments, time, etc. This book will give them the opportunity to relate theory and practicals for a better understanding of the subject.

Comprehensive Practical Chemistry XII Dr. N . K. Verma, B. K. Vermani, Dr. Neera Verma, K . K. Rehani, 2011-11

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Chemistry Laboratory Manual Reena, Indu Bala Sharma, Seema Raj,

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